Appl. No. 10/559,878

Amdt. Dated November 6, 2009

Reply to Office Action of July 6, 2009

## Amendments to the Claims

This listing of claims will replace all prior versions, and listings, of claims in the application:

## Listing of Claims:

 (Currently Amended) A method for the production of an aromatic or hetroaromatic fluorine-labelled compound comprising fluoridation of an iodonium salt of Formula (I) or (ID:

wherein:

Q is an electron deficient aromatic or heteroaromatic moiety;

each of  $R^1$ ,  $R^2$ ,  $R^3$ ,  $R^4$  and  $R^5$  is independently hydrogen, -O( $C_{1\text{-}10}$  alkyl) or  $C_{1\text{-}10}$  alkyl; and

 $\vec{Y}$  is a counter ion such as trifluoromethane sulfonate (triflate), perfluoro  $C_2$ - $C_{10}$  alkyl sulphonate, trifluoroacetate, methane sulfonate (mesylate), toluene sulfonate. (tosylate), tetraphenylborate;

to give a product of general formula (III):

where Q is substituted with one or more substituents selected from  $C_{1-10}$  alkyl,  $-O(C_{1-10}$  alkyl, -C(=O)  $C_{1-10}$  alkyl, -C(=O)  $C_{1-10}$  alkyl, -C(=O)  $C_{1-10}$  alkyl,  $-C(-C_{1-10}$  alkyl),  $-C(-C_{1-10}$  alk

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 $-C(=O)C_{5\text{-}14} \text{ heteroaryl}, -C(=O)NR^6(C_{5\text{-}14} \text{ heteroaryl}), C_{3\text{-}10} \text{ cycloalkyl}, -O(C_{3\text{-}10} \text{ cycloalkyl}), -O(C_{3\text{-}10} \text{ cycl$ 

$$\begin{split} &C(=O)(C_{3\text{-}10} \quad \text{cycloalkyl)}, \quad -C(=O)NR^6(C_{3\text{-}10} \quad \text{cycloalkyl)}, \quad C_{3\text{-}10} \quad \text{heterocyclyl}, \quad -O(C_{3\text{-}10} \quad \text{heterocyclyl}), \quad -C(=O)NR^6(C_{5\text{-}14} \quad \text{heterocyclyl}), \quad \frac{\text{wherein, when $Q$ is}}{C_{5\text{-}14} \quad \text{heterocyclyl}} \\ &\frac{\text{wherein, when $Q$ is}}{C_{5\text{-}10} \quad \text{heterocyclyl}} \\ &\frac{\text{wherein, wh$$

substituted with an electron donating substituent, Q also contains one or more electron

withdrawing groups to ensure Q is electron deficient;

and wherein said fluoridation is carried out with a fluoride ion source characterised in that the reaction solvent is either 100% water or a mixture of water and a water miscible solvent.

(Cancelled)

(Cancelled)

- (Previously presented) A method as claimed in claim 1, wherein the water miscible solvent is acetonitrile, ethanol, methanol, tetrahydrofuran or dimethylformamide.
- (Previously Presented) A method as claimed in claim 1 wherein the volume:volume ratio of water:water-miscible solvent is between 1:99 and 1:1.
- (Original) A method as claimed in claim 5 wherein the volume:volume ratio of water:water-miscible solvent is from 10:90 to 30:70.
- (Previously Presented) A method as claimed in claim 1, wherein the fluoride ion source is potassium, caesium or sodium fluoride.
- (Cancelled)
- 9. (Previously Presented) A method as claimed in claim 1, wherein each of  $R^1$ - $R^5$  is independently selected from hydrogen,  $C_{1-3}$  alkyl and -O-( $C_1$ - $C_3$  alkyl).

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10. (Previously Presented) A method as claimed in claim 1 wherein, in the compound of Formula II, the "solid support" is polystyrene, polyacrylamide, polypropylene or glass or silicon coated with such a polymer.

- 11. (Previously Presented) A method as claimed in claim 1 wherein the solid support is in the form of small discrete particles or is a coating on the inner surface of a reaction vessel.
- 12. (Previously Presented) A method as claimed in claim 1, wherein, in the compound of Formula II the "linker" is C<sub>1-20</sub> alkyl or C<sub>1-20</sub> alkoxy, attached to the resin by an amide ether or a sulphonamide bond or a polvethylene glycol (PEG) linker.
- (Previously Presented) A method as claimed in claim 1
  wherein R<sup>6</sup> is H, C<sub>1</sub>-C<sub>6</sub> alkyl, C<sub>3</sub>-C<sub>10</sub> cycloalkyl, C<sub>3</sub>-C<sub>10</sub> heterocyclyl, C<sub>4</sub>-C<sub>10</sub> aryl or
  C<sub>4</sub>-C<sub>10</sub> heteroaryl:

any of which may optionally be substituted with OH, NHR<sup>6</sup>, COOH or protected versions any of these groups; or alternatively

any two adjacent substituents may form a four- to six-membered carbocyclic or heterocyclic ring, optionally fused to a further aromatic, heteroaromatic, carbocyclic or heterocyclic ring.

- 14. (Previously Presented) A method as claimed in claim 1, wherein the aromatic moiety Q has an additional substituent selected from OH, NHR<sup>6</sup> or halogen.
- 15. (Previously Presented) A method as claimed in claim 1, wherein the group Q is one of the following:

- (Previously Presented) A method as claimed in claim 1, wherein the fluorine-labelled compound is an [18F]-labelled compound and the fluoride ion source is a source of 18F.
- 17. (Previously Presented) A method as claimed in claim 1, wherein the F-labelled compound is selected from the following:

- 18. (Previously Presented) A method as claimed in claim 1, further including, in any order, one or more of the following steps: removal of excess <sup>18</sup>F, for example by ion-exchange chromatography; and/or
  - (i) removal of the protecting groups; and/or
  - (ii) removal of organic solvent; and/or
  - (iii) formulation of the resultant compound as an aqueous solution.
- 19. (Original) A kit for the production of an aromatic fluorine-labelled compound, the kit comprising:
- (i) a vial containing an aqueous solvent for dissolving the fluoride ion source; and
- (ii) a reaction vessel containing an iodonium salt.

20. (Original) A kit as claimed in claim 19, wherein the solvent is 100% water.

21. (Original) A kit as claimed in claim 19 wherein the solvent is a mixture of water and

a water miscible solvent.

22. (Original) A kit as claimed in claim 21, wherein the water miscible solvent is

acetonitrile, ethanol, methanol, tetrahydrofuran or dimethylformamide.

23. (Previously Presented) A kit as claimed in claim 21 wherein the volume:volume ratio

of water:water-miscible solvent is between 1:99 and 1:1.

24. (Original) A kit as claimed in claim 23 wherein the volume:volume ratio of

water:water-miscible solvent is from 10:90 to 30:70.

25. (Previously Presented) A kit as claimed in claim 19 wherein the iodonium salt is

compound of general formula (I) or (II).

26. (Previously Presented) A kit as claimed in claim 20 wherein the iodonium salt is a

compound of general formula (II) and the solid support comprises a coating on the surface of

the reaction vessel.

27. (Previously Presented) A kit as claimed in claim 19, wherein the reaction vessel is a

cartridge or a microfabricated vessel.

28. (Previously Presented) A kit as claimed in claim 19, further comprising a source of

fluoride ions.

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